

# **CHEMISTRY**

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### Supporting Information

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#### **Novel and Effective Copper–Aluminum Propane Dehydrogenation Catalysts**

**Jana Schäferhans,<sup>[b]</sup> Santiago Gómez-Quero,<sup>[a]</sup> Daria V. Andreeva,<sup>\*,[b]</sup> and  
Gadi Rothenberg<sup>\*,[a]</sup>**

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## Experimental Section

### *Materials and Instrumentation*

Transmission and scanning electron microscopy (TEM, Zeiss 922 EFTEM operating at 200 kV and LEO 1530 FE-SEM, Zeiss, respectively), in combination with an ultra microtome (Ultracut E Reichert Jung, thickness 50 nm) were applied to characterize the optical response, structure, and size of the Al-Cu powder.

Powder X-ray diffraction (PXRD) diagrams were acquired using a Stoe STADI P X-ray Transmission diffractometer: Cu  $K\alpha_1$ , irradiation at ambient temperature, with  $2\theta = 5-90^\circ$ .

Surface area and pore size distribution based on physisorption (adsorption and desorption of gases) were measured by the BET<sup>[1]</sup> and BJH<sup>[2]</sup> methods using  $N_2$  at 77 K on the vacuum gas sorption Surfer (Thermo Scientific). All samples were dried under vacuum for 24 h at 300°C prior to each measurement.

Al shots, irregular, 15 mm and down, 99.9% (metals basis) from Alfa Aesar and Cu beads, 2-8 mm, 99.9995% (trace metals basis) from Sigma Aldrich were used as received. The water was purified before use in a three stage Millipore Milli-Q Plus 185 purification system and had a resistivity higher than 18.2 M $\Omega$ ·cm. All gases were supplied by Praxair (=99.995% pure). The  $H_2$ , Ar and  $O_2$  streams were further purified over molecular sieves and/or BTS columns at 80°C prior to use.

### *Procedure for synthesising the Pt-Sn/Al<sub>2</sub>O<sub>3</sub> reference catalyst*

The bimetallic Pt-Sn/Al<sub>2</sub>O<sub>3</sub> was prepared by simultaneous addition of the precursor solutions (1 mol% loading of each metal). First, 13.0 cm<sup>3</sup> of an acidified 15 mM SnCl<sub>2</sub>·2H<sub>2</sub>O solution was added to 59.0 cm<sup>3</sup> of a 3 mM H<sub>2</sub>PtCl<sub>6</sub>·6H<sub>2</sub>O solution. Then, 2.0 g of support were added and the volume was adjusted to 200 cm<sup>3</sup>. The resulting slurry was heated to dryness at 80°C (2°C min<sup>-1</sup>) under an Ar purge. The solid residue was dried overnight in an Ar stream (60 cm<sup>3</sup> min<sup>-1</sup>) at 80°C (heating ramp of 2°C min<sup>-1</sup>), then cooled to room temperature and stored in a desiccator in the dark. Prior to catalysis, all catalyst samples were ground and sieved to <200  $\mu$ m and activated in 80 cm<sup>3</sup> min<sup>-1</sup> 25% v/v H<sub>2</sub>/Ar to 600°C (5°C min<sup>-1</sup>) for 1 h.

## References

- [1] S. Brunauer, L. S. Deming, W. E. Deming, E. Teller, *J. Am. Chem. Soc.* **1940**, 62, 1723-1732.
- [2] E. P. Barrett, L. G. Joyner, P. P. Halenda, *J. Am. Chem. Soc.* **1951**, 73, 373-380.